P510/1 PHYSICS Paper 1 Nov./Dec. 2025 2½ hours



# UGANDA NATIONAL EXAMINATIONS BOARD

# **Uganda Advanced Certificate of Education**

## **PHYSICS**

Paper 1 (Theory)

2 hours 30 minutes

#### INSTRUCTIONS TO CANDIDATES:

This paper consists of three Sections; A, B and C.

Answer five questions in all including at least one, but not more than two from each of the sections.

Any additional question(s) answered will not be marked.

Begin answering each question on a fresh page.

Silent, non-programmable scientific calculators may be used.

Assume where necessary:

Acceleration due to gravity, g =  $9.81 \text{ m s}^{-2}$ Electron charge, e =  $1.6 \times 10^{-19} \text{ C}$ Electron mass =  $9.11 \times 10^{-31} \text{ kg}$ Mass of the earth =  $5.97 \times 10^{24} \text{ kg}$ Plank's constant, h =  $6.6 \times 10^{-34} \text{ J s}$ 

Stefan's – Boltzmann's constant,  $\sigma = 5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$ 

Radius of the earth  $= 6.4 \times 10^{6} \, \mathrm{m}$ Radius of the sun  $= 7.0 \times 10^{8} \, \mathrm{m}$ Radius of the earth's orbit about the sun  $= 1.5 \times 10^{11} \, \mathrm{m}$ Speed of light in a vacuum, c  $= 3.0 \times 10^{8} \, \mathrm{m s^{-1}}$ Thermal conductivity of copper  $= 390 \, \mathrm{W m^{-1} \, K^{-1}}$ Thermal conductivity of aluminium  $= 210 \, \mathrm{W \, m^{-1} \, K^{-1}}$ Specific heat capacity of water  $= 4,200 \, \mathrm{J \, kg^{-1} \, K^{-1}}$ 

Specific heat capacity of water  $= 4,200 \text{ J kg}^{-1} \text{ K}^{-1}$ Universal gravitational constant,  $G = 6.67 \times 10^{-11} \text{ N m}^2 \text{ kg}^{-2}$ Surface tension of water  $= 7.0 \times 10^{-2} \text{ N m}^{-1}$ 

Density of water  $= 1,000 \text{ kg m}^3$ Gas constant, R  $= 8.31 \text{ J mol}^{-1} \text{ K}^{-1}$ Charge to mass ratio, e/m  $= 1.8 \times 10^{11} \text{ C kg}^{-1}$ Avogadro's number,  $N_A$   $= 6.02 \times 10^{23} \text{ mol}^{-1}$ 

Turn Over



## SECTION A

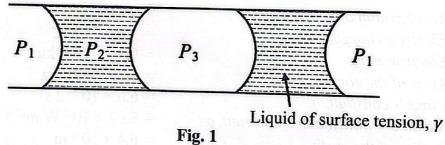
1. (a) Define surface tension in terms of;

- (i) force. (01 mark)
- (ii) work. (01 mark)
- (b) (i) Derive an expression for excess pressure inside a soap bubble of radius, r. (04 marks)
  - (ii) Calculate the total pressure inside an air bubble of radius  $1.0 \times 10^{-5}$  m at a depth of 0.3 m below the surface of water. (Atmospheric pressure is  $1.0 \times 10^{5}$  N m<sup>-2</sup>.) (03 marks)
  - (iii) Find the height to which the mercury rises if the bubble in (b) (ii) is introduced into a manometer containing mercury.

    (02 marks)
- (c) Two soap bubbles of radii,  $r_1$  and  $r_2$  are brought together so that the combined bubble has a common interface of radius,  $R_T$ . Given that  $r_1 < r_2$ , show that the radius of the interface is given by:

$$R_T = \frac{r_1 r_2}{r_2 - r_1} \tag{05 marks}$$

(d) Figure 1 shows sections of a capillary tube at pressures  $P_1$ ,  $P_2$  and  $P_3$  as indicated in the figure 1:



If the radius of the tube is r, derive an expression for  $P_3$  in terms of  $P_1$ .

(04 marks)

- 2. (a) Derive the principle of conservation of energy using Newton's laws of motion. (06 marks)
  - (b) A particle of mass 5 g moving with a velocity of 10 m s<sup>-1</sup> in a straight line strikes a stationery particle of mass 4 g. If the two particles collide inelastically, find their final velocity.

    (02 marks)

Two balls, A and B of masses  $m_1$  and  $m_2$ , initially approaching (c) (i) each other with velocities  $u_1$  and  $u_2$  respectively, had a head-on collision. A continued in its original direction with velocity  $v_1$ and B moved opposite to its original direction with a velocity  $v_2$ . If the collision is perfectly elastic, show that: (05 marks)

 $u_1 + u_2 = v_1 + v_2$ .

- Given that the mass of A is 2 kg and the mass of B is 3 kg, (ii) initially moving with velocities of 15 m s<sup>-1</sup> and 20 m s<sup>-1</sup> respectively, find the final velocity of A if the final velocity of B(02 marks) is 10 m s<sup>-2</sup>.
- State any one application of inelastic collision. (01 mark)
- Explain the motion of a rocket using the principle of conservation of (d) (04 marks) momentum.
- (01 mark) Define Simple Harmonic Motion (S.H.M). 3. (a)
  - Sketch the graphs for a body performing S.H.M for the following; (b) (02 marks) velocity against displacement. (i)
    - (02 marks) (ii) mechanical energy against displacement.
  - A small bob of mass 0.2 kg is suspended by an inextensible string (c) of length 0.8 m. If the bob is whirled in a horizontal circle of radius 0.4 m, find the;
    - (03 marks) angular speed of the bob. (i)
    - (02 marks)tension in the string. (ii)
  - State the law of floatation. (01 mark)(i) (d)
    - An object is immersed in a fluid. Show that the weight of the (ii) fluid displaced is equal to the upthrust exerted by the fluid on the object. (04 marks)
  - A cylindrical vessel of cross-sectional area A, contains air of volume V (e) at a pressure P trapped by a frictionless airtight piston of mass M. If the piston is pushed down and released;
    - show that the resulting motion is simple harmonic. (i)
    - (ii) find the frequency of the piston. (05 marks)

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- 4. (a) State Kepler's laws of planetary motion. (03 mark)
  - (b) Derive Kepler's third law using Newton's law of gravitation. (03 marks)
  - (c) Describe the circumstances under which a body is said to be in a state of weightlessness. (04 marks)
  - (d) What is meant by **uniform velocity**? (01 mark)
  - (e) A bus travelling at a uniform velocity of 30 m s<sup>-1</sup> passes a stationary car, which begins to move 10 s later with uniform acceleration of 4 m s<sup>-2</sup> in the same direction as the bus. How far does the car travel before it catches up with the bus?

    (04 marks)
  - (f) A stone is projected from a level ground with a speed u, at an angle  $\theta$  to the horizontal. Derive an expression for the horizontal distance the stone travels to the point of landing on the ground. (05 marks)

# **SECTION B**

5. (a) (i) Define thermal conductivity. (01 mark) (ii) Explain the mechanism of heat conduction in an insulator.

(03 marks)

(b) (i) With the aid of a labelled diagram, describe an experiment to determine the thermal conductivity of a poor conductor.

(08 marks)

- (c) (i) Explain how a black body may be approximated. (03 marks)
  - (ii) Solar radiation is incident on a planet whose distance from the sun is  $2.3 \times 10^{11}$  m. If the sun is a sphere of radius  $7.0 \times 10^8$  m and its temperature is 6,000 K, calculate the temperature of the surface of the planet. (05 marks)
- 6. (a) (i) State any two thermometric properties used in gas thermometers.

  (01 mark)
  - (ii) Why are different values of temperature obtained when a platinum resistance and liquid-in-glass thermometers are used to measure the same temperature of a liquid? (01 mark)
  - (b) In a constant-volume gas thermometer, the difference in the levels of mercury is -100 mm at melting pure ice, 140 mm at the temperature of steam at standard atmospheric pressure, and 30 mm at unknown temperature of a liquid. Calculate the temperature of the liquid.

    (02 marks)

- (c) Define the following;
  - (i) specific heat capacity,
  - (ii) specific latent heat of vaporization, of a substance. (02 marks)
- (d) (i) Describe, with the aid of a labelled diagram, the continuous-flow method for determining the specific heat capacity of a liquid.

  (05 marks)
  - (ii) When a potential difference (p.d) across a coil immersed in water is 5.25 V and the current through it is 2.58 A, a mass of 6.85 g of water evaporates in 20 minutes. Suppose a mass of 2.38 g of water evaporates in the same time duration when the p.d and current are maintained at 3.20 V and 1.57 A respectively, what is the specific latent heat of vaporisation of water? (05 marks)
- (e) Explain the occurrence of;
  - (i) sea breeze, (02 marks) (ii) land breeze. (02 marks)
- 7. (a) (i) State Boyle's law. (01 mark)
  (ii) Using a labelled diagram, describe an experiment to verify
  - Using a labelled diagram, describe an experiment to verify Boyle's law. (05 marks)
  - (b) (i) State four differences between real and ideal gases. (04 marks)
    - (ii) Using the same axes, sketch graphs of pressure against volume for an ideal gas at three different temperatures. (02 marks)
  - (c) (i) Define saturated vapour pressure. (01 mark)
    - (ii) A sealed container at a temperature of 27 °C contains air and saturated water vapour. The total pressure inside the container is 1.01 × 10<sup>5</sup> Pa. If the saturated vapour pressure of water at 27 °C is 2.0 × 10<sup>3</sup> Pa and at 80 °C is 3.51 × 10<sup>3</sup> Pa, calculate the total pressure inside the container at 80 °C, assuming the vapour pressure remains saturated. (04 marks)
  - (d) The root mean square speed of molecules of one mole of a gas is  $44.72 \text{ m s}^{-1}$ . Find the temperature of the gas if its density and volume are  $9.0 \times 10^{-2} \text{ kg m}^{-3}$  and  $42.0 \text{ m}^3$  respectively. (03 marks)

# SECTION C

8.	(a)	(i)	What is meant by cathode rays?	(01 mark)	
(7 A	(4)	(ii)	Distinguish between cathode rays and gamma rays.	(02 marks)	
		(iii)	Explain how cathode rays are produced.	(04 marks)	
	(b)	With	the aid of a labelled diagram, describe an experiment	experiment to	
	a ni la	determine specific charge of an electron.		(06 marks)	
	(c)	An electron with energy of 480 eV moves at right angles to a uniform magnetic field of flux density $1.5 \times 10^{-3}$ T:			
		(i)	Show that the path of the electron is circular.	(03 marks)	
		(ii) Find the radius of the circular path followed by the el			
				(04 marks)	
9.	(a)	Define the following;			
		(i)	threshold frequency.	(01 mark)	
		(ii)	work function.	(01 mark)	
	(b)	Ultra violet (UV) light is directed onto a cleaned Zinc plate connected			
		to the cap of a gold leaf electroscope. Explain the observations made when the gold leaf electroscope is;			
		(i)	negatively charged,	(03 marks)	
		(ii)	positively charged.	(03 marks)	
	(c)				
	om Mil)	photoelectrons of maximum velocity $6.0 \times 10^5$ m s <sup>-1</sup> are emitted. Calculate the;			
		(i)	work function for the metal.	(04 marks)	
		(ii)	threshold frequency.	(02 marks)	
	(d)	(i)	Sketch a graph to show a typical X-ray spectrum.	(02 marks)	
	odt bin	(ii)	Explain the main features of the graph in (d) (i).	(04 marks)	
10.	(a)	(i)	Define radioactivity.	(01 mark)	
	372	(ii)	With the aid of a labelled diagram, describe how a d		
		, victus	cloud chamber is used to detect the type of radioactive radiations (06 marks)		
	(b)	(i)	(i) Sketch a graph showing how current varies with applied voltage		
	( )		in an ionisation chamber.	(01 mark)	
		(ii) Explain the main features of the sketched grap		b) (i).	
				(03 marks)	

- (iii) A source of  $\alpha$ -particles has an initial activity of  $2 \times 10^5$  disintegrations per second. When the  $\alpha$ -particles enter an ionisation chamber, a saturation current of  $2.0 \times 10^{-7}$ A is obtained. If the energy required to produce an ion pair is 32 eV, find the energy of one  $\alpha$ -particle. (04 marks)
- (c) (i) State four uses of radio isotopes. (02 marks)
  - (ii) A small amount of radio isotope <sup>24</sup>Na is smeared on to a card and its activity falls by 87.5% in 45 hours. Find the half-life of <sup>24</sup>Na. (03 marks)

**END** 

